J. Chem. Research (S), 1998, 164–165†

Novel *N*-Substituted Amino-4-methylsulfanyl-2pyridones and Deazapurine Analogues from Ketene Dithioacetals†

Galal H. Elgemeie, *a Ahmed H. Elghandour, b Ali M. Elzanate b and Wafaa A. Masoud b

^aChemistry Department, Faculty of Science, Helwan University, Cairo, Helwan, Egypt ^bChemistry Department, Faculty of Science, Cairo University (bani Suef Branch), Bani Suef, Egypt

A novel synthesis of *N*-substituted amino-4-methylsulfanyl-2-pyridones and deazapurine analogues *via* the reaction of ketene dithioacetals with substituted semi- and thio-semicarbazide derivatives is reported and the synthetic potential of the method is demonstrated.

Synthetic analogues of purines are widely used in the medical sciences and in clinical medicine. Examples include the 6-sulfanylguanine and 6-sulfanylpurine which are widely used clinically. The purine analogue 4-hydroxypyrazolopyrimidine (allopurine), used in the treatment of hyperuricemia and gout, inhibits de novo purine biosynthesis and xanthine oxidase. Azathiopurine, which is catabolized to 6-sulfanylpurine, is employed in organ transplantation to supress events involved in immunologic rejection. As a part of our program directed towards the development of new simple and efficient procedures for the synthesis of antimetabolities, we have recently reported different successful approaches for the synthesis of sulfanylpurine, 5-deazafolic acid and deazapyrimidine nucleosides.² Derivatives of these ring systems are interesting because they have useful properties as antimetabolites in biochemical reactions.⁴ The present research deals with a novel synthesis of N-substituted amino-4-methylsulfanyl-2-pyridones and deazapurine analogues using ketene dithioacetals. Thus, it has been found that compounds 1 reacted with 4-substituted 1-cyanoacetylthiosemicarbazide 3a,b at room temperature in the presence of pulverized potassium hydroxide in 1,4dioxane to give the corresponding N-(4-methylsulfanyl-2oxo-1-pyridyl)thiourea derivatives 4. The structures of compounds 4 were established on the basis of their elemental analysis and spectral data (MS, ¹H NMR, ¹³C NMR and IR). Thus, structure 4a is supported by its mass spectrum which showed a molecular ion corresponding to the formula $C_{15}H_{12}N_6S_2O$ (M⁺ = 356). The ¹H NMR spectrum revealed a band at δ 2.78 assignable to the SCH₃ group, a multiplet at δ 7.21–7.60 assigned to aromatic protons, a broad singlet at δ 8.78 assignable to an amino group and two broad singlets at δ 9.85 and 10.69 assigned to the NH protons. The ¹³C NMR spectrum was characterized by a signal at δ 19.81 attributed to the SCH₃ carbon and two signals at δ 113.17 and 115.52 attributed to the two CN carbons. Moreover, signals appeared at δ 124.32, 127.56, 146.34, 149.12 and 161.93 corresponding to C-5, C-3, C-4, C-6 and C-2, respectively.

The formation of **4** from the reaction of **1** with **3** is assumed to proceed *via* Michael addition of the active methylene of **3** to the double bond in **1**. The formed Michael adducts then cyclized smoothly *via* MeSH elimination and addition to the cyano group. In a typical experiment, when the ketene dithioacetals **1a,b** reacted with cyanoacetohydrazide **2** at room temperature in the presence of KOH–1,4-dioxane, the *N*-amino-4-methylsulfanyl-2-

pyridones **5a,b** were obtained in good yield. In related work, Peseke *et al.*⁵ has reported the synthesis of compound **6** by the reaction of **1** with cyanoacetohydrazide **2** in unreported conditions. The structures of **5** were established and confirmed on the basis of their elemental analysis and spectral data (MS, 1 H NMR, 13 C NMR and IR). The analytical data for **5a** revealed a molecular formula $C_8H_7N_5SO$ (M $^{+}$ = 221) and 1 H NMR spectroscopy was used to confirm

Scheme 1

^{*}To receive any correspondence.

[†]This is a **Short Paper** as defined in the Instructions for Authors, Section 5.0 [see *J. Chem. Research (S)*, 1998, Issue 1]; there is therefore no corresponding material in *J. Chem. Research (M)*.

the structure. Thus, ${}^{1}H$ NMR revealed a band at δ 2.68 assignable to the SMe group and two broad singlets at δ 5.52 and 8.42 assignable to two amino groups. The 13 C NMR spectrum revealed a signal at δ 18.93 assigned to the SMe group and signals appeared at δ 120.32, 122.90, 147.14, 149.17 and 160.96 corresponding to C-5, C-3, C-4, C-6 and C-2, respectively. The formation of 5 from the reaction of 1 and cyanoacetohydrazide 2 is assumed to proceed via the intermediacy of Michael adducts, which cyclized to yield the final N-amino-2-pyridones. The reaction of ketene dithioacetals with Schiff bases was also examined. Thus, when **1a.b** were reacted with 1-cyanoacetyl-4-arylmethylidenesemicarbazide 6 in the presence of KOH-1,4-dioxane, the 2pyridone-N-Schiff bases 8 were obtained. The structures of 8 were established on the basis of elemental analysis and spectral data (MS, ¹H NMR, ¹³C NMR and IR). The analytical data for 8c revealed a molecular formula C16H13N5SO (M⁺ = 323). The ¹³C NMR showed a signal at δ 18.15 due to the SMe carbon and a signal at δ 160.17 attributed to a 2-pyridone carbonyl carbon. Compounds 4 and 8 can also be prepared by the reaction of the corresponding Namino-2-pyridones 5 with substituted isothiocyanates and aldehydes, respectively, in refluxing 1,4-dioxane for 2h. Compounds 8 reacted with hydrazine in refluxing ethanol to give the corresponding pyrazolo[3,4-c]pyridines 9. The structures of each of the compounds 9 were established on the basis of elemental analysis and spectral data.

In summary, we have achieved a regiospecific synthesis of interesting N-substituted amino-4-methylsulfanyl-2pyridones and deazapurine analogues via the reaction of ketene dithioacetals with semi- and thio-semicarbazidederivatives. The products obtained are currently under biological evaluation studies.

Experimental

All melting points are uncorrected. IR spectra were obtained (KBr) on a Pye Unicam instrument. $^1\text{H-}$ and $^{13}\text{C-NMR}$ spectra were measured on a Varian 400 or Wilmad 270 MHz spectrometer for (CD₃)₂SO solutions using SiMe₄ as internal standard. Mass spectra were recorded on a Varian MAT 112 spectrometer. Analytical data were obtained from the Microanalytical Data Centre at Cairo University.

N-(4-Methylsulfanyl-2-oxo-1-pyridyl)thiourea Method A.—A mixture of [bis(methylsulfanyl)methylene]malono $nitriles \quad \textbf{1a} \quad \text{or} \quad ethyl \quad 2\text{-cyano-3,3-bis} (methylsulfanyl) acrylate \quad \textbf{1b}$ (0.01 mol),4-substituted cvanoacetythiosemicarbazide (0.01 mol), potassium hydroxide (0.012 mol) and dry 1,4-dioxane (50 ml) were stirred at room temperature for 24 h. The reaction mixture was acidified with hydrochloric acid and the formed precipitate was collected by filtration, dried and then crystallized from the appropriate solvent.

Method B.—To a solution of N-amino-2-pyridones 5 (0.01 mol) in 1,4-dioxane (50 ml), phenyl isothiocyanate or benzoyl isothiocyanate (0.01 mol) was added. The resulting mixture was refluxed for 2h and the solid product collected by filtration and crystallized from the appropriate solvent.

4a: yield 52%, mp 280–282 °C (from EtOH); $\nu_{\rm max}$ (KBr)/cm⁻¹ 3391 (NH, NH₂), 2217 (CN), 1655 (CO) (Found: C, 50.3; H, 3.6; N, 23.5%. C₁₅H₁₂N₆S₂O requires C, 50.7; H, 3.4; N, 23.6%).

4b: yield 52%, mp 242–244 °C (from 1,4-dioxane); ν_{max} (KBr)/cm⁻¹ 3600, 3380, 3320–3100 (OH, NH), 2212 (CN), 1655 (CO) (Found: C, 50.7; H, 3.0; N, 19.4%. C₁₅H₁₁N₅S₂O₂ requires C, 50.4; H, 3.1; N, 19.6%).

4c: yield 53%, mp > 300 °C (from EtOH); ν_{max} (KBr)/cm⁻¹ 3312, 3280 (NH, NH₂, 2211 (CN), 1650 (CO); $\delta_{\rm H}$ (DMSO) 2.69 (s, 3 H, SMe), 6.92 (m, 2 H, NH₂), 7.19–7.77 (m, 5 H, Ph), 9.60 (s, br, 1 H, NH), 10.53 (s, br, 1 H, NH) (Found: C, 49.7; H, 3.2; N, 21.9%. $C_{16}H_{13}N_6S_2O_2$ requires C, 50.0; H, 3.0; N, 21.6%).

4d: yield 55%, mp 290–293 °C (from MeOH); ν_{max} (KBr)/cm⁻¹ 3454, 3350 (OH, NH), 2213 (CN), 1634 (CO). 2.82 (s, 3 H, SMe), 7.18-7.81 (m, 5 H, Ph), 11.35 (s, br, 1 H, NH), 13.50 (s, br, 1 H, NH) (Found: C, 49.5; H, 3.0; N, 18.4%. C₁₆H₁₁N₅S₂O₃ requires C, 49.9; H, 2.9; N, 18.2%).

N-Amino-4-methylsulfanyl-2-pyridone Derivatives 5a,b—General Procedure.—A mixture of [bis(methylsulfanyl)methylene]malononitriles 1a or ethyl 2-cyano-3,3-bis(methylsulfanyl)acrylate 1b (0.01 mol), cyanoacetohydrazide 2 (0.01 mol), and potassium hydroxide (0.012 mol) in dry 1,4-dioxane (50 ml) was stirred at room temperature for 24 h. The reaction mixture was acidified with hydrochloric acid and the formed precipitate was collected by filtration, dried and then recrystallized from the appropriate solvent.

5a: yield 40%, mp $> 300 \,^{\circ}\text{C}$ (from MeOH), ν_{max} (KBr)/cm⁻¹ **Sat.** yield 40%, hip > 300 C (from MeOH), ν_{max} (KBI)/Cli 3549, 3292 (NH₂), 2216 (CN), 1734 (CO) (Found: C, 43.6; H, 3.3; N, 31.5%. $C_8H_7N_5SO$ requires C, 43.4; H, 3.2; N, 31.6%). **5b.** yield 35%; mp 150–151 °C (from MeOH); ν_{max} (KBr)/cm⁻¹

3609, 3316 (OH, NH₂), 2213 (CN), 1734 (CO) (Found: C, 43.4; H, 2.9; N, 25.0%. C₈H₆N₄SO₂ requires C, 43.2; H, 2.7; N, 25.5%).

1-(N-Substituted)arylmethylideneamino-4-methylsulfanyl-2-pyridone Derivatives 8a-f. Method A.—A mixture of [bis(methylsulfanyl)-methylene]malononitriles 1a or ethyl 2-cyano-3,3-bis(methylsulfanyl)acrylate 1b (0.01 mol), 1-cyanoacetyl-4-arylidenesemicarbazide (0.01 mol), potassium hydroxide (0.012 mol) and 1,4-dioxane (50 ml) were stirred at room temperature for 24 h. The reaction mixture was acidified with hydrochloric acid and the precipitate formed was collected by filtration, dried and then recrystallized from the appropriate solvent.

Method \hat{B} . To a solution of N-amino-2-pyridones 5 (0.01 mol) in 1,4-dioxane (50 ml), aromatic aldehyde (0.01 mol) was added. The resulting mixture was refluxed for 2h and the solid product

collected by filtration and crystallized from the appropriate solvent. **8a**: yield 51%; mp > 300 °C (from 1,4-dioxane); ν_{max} (KBr)/cm⁻¹ 3425 (NH₂), 2211 (CN), 1634 (CO) (Found: C, 52.7; H, 3.1; N, 20.2%. C₁₅H₁₀ClN₅SO requires C, 52.4; H, 2.9; N, 20.4%).

8b: yield 83%; mp > 300 (from 1,4-dioxane); ν_{max} (KBr)/cm⁻¹ 3446 (NH₂), 2220 (CN), 1685 (CO) (Found: C, 59.0; H, 4.2; N, 21.5%. C₁₆H₁₃N₅SO requires C, 59.4; H, 4.0; N, 21.7%).

8c: yield 76%; mp 285–287°C (from MeOH); $\nu_{\rm max}({\rm KBr})/{\rm cm}^{-1}$ 3506–3345 (OH), 2209 (CN), 1654 (CO) (Found: C, 58.2; H, 3.3; N,

17.8%. $C_{15}H_{10}N_4SO_2$ requires C, 58.0; H, 3.2; N, 18.0%). **8d**: yield 62%; mp 210–211 °C (from EtOH); v_{max}/cm^{-1} 3498, 3381 (NH₂), 2206 (CN), 1687 (CO) (Found: C, 59.0; H, 3.9; N, 17.3%. C₁₆H₁₂N₄SO₂ requires C, 59.3; H, 3.7; N, 17.3%).

6-Amino-3-cyanopyrazolo[3,4-c]pyridin-2(1H)-one Derivatives 9a-d. General Procedure.—A mixture of equivalent amounts of 5b,c,e,f (0.01 mol) and hydrazine hydrate (0.01 mol) was heated in ethanol (30 ml) for 4h. The solid product formed was collected by filtration and crystallized from the appropriate solvent.

9a: yield 53%; mp > 300 °C (from DMF); ν_{max} (KBr)/cm⁻¹ 3479, 3315 (NH, NH₂), 2210 (CN), 1655 (CO); $\delta_{\rm H}$ (DMSO) 5.68 (s, br, 2 H, NH₂), 6.87-7.89 (m, 4 H, C₆H₄), 8.12 (s, 1 H, ylidic CH), 8.28 (s, br, 2 H, NH₂), 11.85 (s, br, 1 H, NH) (Found: C, 51.5; H, 2.9; N, 30.2%. C₁₄H₁₀ClN₇O requires C, 51.3; H, 3.1; N, 29.9%).

9b: yield 55%; mp > 300 °C (from EtOH); ν_{max} (KBr)/cm⁻¹ 3600–3182 (NH, NH₂), 2206 (CN), 1639 (CO) (Found: C, 58.2; H, 4.3; N, 32.2%. C₁₅H₁₃N₇O requires C, 58.6; H, 4.2; N, 31.9%).

9c: yield 54%; mp > 300 °C (from 1,4-dioxane); ν_{max} (KBr)/cm⁻¹ 3450, 3350 (OH, NH, NH₂), 2205 (CN), 1702 (CO) (Found: C, 57.4; H, 3.5; N, 28.3%. C₁₄H₁₀N₆O₂ requires C, 57.1; H, 3.4; N,

9d: yield 54%; mp > 300 °C (from MeOH); ν_{max} (KBr)/cm⁻¹ 3451, 3352 (OH, NH, NH₂), 2205 (CN), 1703 (CO) (Found: C, 58.0; H, 4.0; N, 27.5%. C₁₅H₁₂N₆O₂ requires C, 58.4; H, 3.9; N,

Received, 18th March 1997; Accepted, 13th August 1997 Paper E/7/01889D

References

- 1 G. H. Elgemeie and B. A. Hussain, Tetrahedron, 1994, 50, 199.
- 2 G. H. Elgemeie, A. M. Attia, D. S. Farag and S. M. Sherif, J. Chem. Soc., Perkin Trans. 1, 1994, 1285.
- 3 G. H. Elgemeie, S. E. El-Ezbawy, H. A. Ali and A. K. Mansour, Bull. Chem. Soc. Jpn, 1994, 67, 738.
- 4 T. Tsukamoto, W. H. Haile, J. J. McGuire and J. K. Coward, J. Med. Chem., 1996, 39, 2536.
- K. Peseke, J. Q. Suarez, F. Napoies and M. Basilia, Ger. (East) DD 294 943 (Cl. C07D 487/04) (Chem. Abstr., 1992, 116, 128960